#### Is my water safe?

We are pleased to present this year's Annual Water Quality Report (Consumer Confidence Report) as required by the Safe Drinking Water Act (SDWA). This report is designed to provide details about where your water comes from, what it contains, and how it compares to standards set by regulatory agencies. This report is a snapshot of last year's water quality. We are committed to providing you with information because informed customers are our best allies.

### Do I need to take special precautions?

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Water Drinking Hotline (800-426-4791).

### Where does my water come from?

Our groundwater is from the cockfield formation

## Source water assessment and its availability

Our source water assessment has been completed and is ranked moderate in terms of susceptibility and contamination. For a copy or questions please call 601-992-9714.

Why are there contaminants in my drinking water?

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the Environmental Protection Agency's (EPA) Safe Drinking Water Hotline (800-426-4791).

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity: microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife; inorganic contaminants, such as salts and metals, which can be naturally occurring or result from urban stormwater runoff, industrial, or domestic wastewater discharges, oil and gas production, mining, or farming; pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses; organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems; and radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities. In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

#### How can I get involved?

The Pearl River Valley Water Supply District has regular board meetings on the third Thursday of each month at 9:30 A.M., or visit the website at TheRez.ms

#### FLOURIDE

To comply with the "Regulation Governing Fluoridation of Community Water Supplies", PRVWSD-MAIN HARBOR is required to report certain results pertaining to fluoridation of water system. The number of months in the previous year calendar year in which average fluoride sample results were within the optimal range of 0.7-1.3 ppm was 12. The percentage of fluoride samples collected in the previous calendar year that was within the optimal range of 0.7-1.3 ppm was 96%

#### Results of radon monitoring

Radon is a radioactive gas that you can't see, taste, or smell. It is found throughout the U.S. Radon can move up through the ground and into a home through cracks and holes in the foundation. Radon can build up to high levels in all types of homes. Radon can also get into indoor air when released from tap water from showering, washing dishes, and other household activities. Compared to radon entering the home through soil, radon entering the home through tap water will in most cases be a small source of radon in indoor air. Radon is a known human carcinogen. Breathing air containing radon can lead to lung cancer. Drinking water containing radon may also cause increased risk of stomach cancer. If you are concerned about radon in your home, test the air in your home. Testing is inexpensive and easy. Fix your home if the level of radon in your air is 4 picocuries per liter of air (pCi/L) or higher. There are simple ways to fix a radon problem that aren't too costly. For additional information, call your state radon program or call EPA's Radon Hotline (800-SOS-RADON).

#### Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. MAIN HARBOR is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

## **Water Quality Data Table**

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of contaminants in water provided by public water systems. The table below lists all of the drinking water contaminants that we detected during the calendar year of this report. Although many more contaminants were tested, only those substances listed below were found in your water. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful in our drinking water. Removing all contaminants would be extremely expensive, and in most cases, would not provide increased protection of public health. A few naturally occurring minerals may actually improve the taste of drinking water and have nutritional value at low levels. Unless otherwise noted, the data presented in this table is from testing done in the calendar year of the report. The EPA or the State requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not vary significantly from year to year, or the system is not considered vulnerable to this type of contamination. As such, some of our data, though representative, may be more than one year old. In this table you will find terms and abbreviations that might not be familiar to you. To help you better understand these terms, we have provided the definitions below the table.

		MCLG	MCL,			, ,				9°	
		1	TT, or		Ra	nge	Sample				
Conta	minan <i>ts</i>	MRDLG	MRDL	Water	Low	High	<u>Date</u>	<b>Violation</b>	Typical Source		
	Disinfectants & Disinfectant By-Products										
					sinfect	ant is n	ecessary:	for control o	f microbial contaminants	)	

hlorine (as Cl2) opm)	4	4	1.4	1	1.5	2013	No	Water additive used to control microbes
THMs [Total rihalomethanes] opb)	NA	80	12	NA		2013	No	By-product of drinking water disinfection
Ialoacetic Acids HAA5) (ppp)	NA	60	2	NA		2013	No	By-product of drinking water chlorination
norganic Contaminat	ts		A			, ,		
Nitrate [measured as Vitrogen] (ppm)	10	10	0.08	0.08	0.08	2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Vitrite [measured as Vitrogen] (ppm)	1	1	0.02	0.02	0.02	2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Antimony (ppb)	6	6	0.5	NA		2012	No	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder; test addition.
Arsenic (ppb)	0	10	0.5	NA		2012	No	Erosion of natural deposits; Runoff from orchards; Runoff from glass and electronics production wastes
Barium (ppm)	2	2	0.00847	NA		2012	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Beryllium (ppb)	4	4	0.5	NA		2012	No	Discharge from metal refineries and coal-burning factories; Discharge from electrical, aerospace, and defense industries
Cadmium (ppb)	5	5	0.5	NA		2012	No	Corrosion of galvanized pipe: Erosion of natural deposits; Discharge from metal refineries; runoff from waste batteries and paints
Chromium (ppb)	100	100	0.5	NA		2012	No	Discharge from steel and pulp mills; Erosion of natural deposits
Cyanide [as Free Cn] (ppb)	200	200	15	NA		2012	No	Discharge from plastic and fertilizer factories; Discharge from steel/metal factories
Fluoride (ppm)	4	4	1.15	NA		2012	No	Erosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer and aluminum factories
Mercury [inorganic] (ppb)	2	2	0.5	NA		2012	No	Erosion of natural deposits; Discharge from refineries an factories; Runoff from landfills; Runoff from cropland

delenium (ppb)	50	50	2.5	NA	2010	No	Discharge from petroleum and metal refineries; Erosion of natural deposits; Discharge from mines
Thallium (ppb)	0.5	2	0.5	NA	2012	No	Discharge from electronics, glass, and Leaching from ore-processing sites; drug factories
olatile Organic Cont	aminants				AND		
Benzene (ppb)	0	5	0.5	NA	2009	No	Discharge from factories; Leaching from gas storage tanks and landfills
Carbon Tetrachloride	0	5	0.5	NA	2009	No	Discharge from chemical plants and other industrial activities
Chlorobenzene monochlorobenzene) ppb)	100	100	0.5	NA	2009	No	Discharge from chemical and agricultural chemical factories
o-Dichlorobenzene (ppb)	600	600	0.5	NA	2009	No	Discharge from industrial chemical factories
p-Dichlorobenzene (ppb)	75	75	0.5	NA	2009	No	Discharge from industrial chemical factories
1,2-Dichloroethane (ppb)	0	5	0,5	NA	2009	No	Discharge from industrial chemical factories
1,1-Dichloroethylene (ppb)	7	7	0.5	NA	2009	No	Discharge from industrial chemical factories
cis-1,2-Dichloroethyl ene (ppb)	70	70	0.5	NA	2009	No	Discharge from industrial chemical factories
trans-1,2-Dichloroeth ylene (ppb)	100	100	0.5	NA	2009	No	Discharge from industrial chemical factories
Dichloromethane (ppb)	0	5	0.5	NA	2009	No	Discharge from pharmaceutical and chemical factories
1,2-Dichloropropane (ppb)	0	5	0.5	NA	2009	No	Discharge from industrial chemical factories
1,2,4-Trichlorobenze ne (ppb)	70	70	0.5	NA	2009	No	Discharge from textile-finishing factories
1,1,1-Trichloroethane (ppb)	200	200	0.5	NA	2009	No	Discharge from metal degreasing sites and other factories
1,1,2-Trichloroethane (ppb)	3	5	0.5	NA	2009	No	Discharge from industrial chemical factories
Ethylbenzene (ppb)	700	700	0.5	NA	2009	No	Discharge from petroleum refineries
Styrene (ppb)	100	100	0.5	NA	2009	No	Discharge from rubber and plastic factories; Leaching from landfills
Tetrachloroethylene (ppb)	0	5	0.5	NA	2009	No	Discharge from factories and dry cleaners

Trichloroethylene (ppb)	0	5	0.5	NA	2009	ľ	Vo de	scharge from metal greasing sites and other ctories
Toluene (ppm)	1	1	0.0005	NA	2009	ı	NA 1	ischarge from petroleum ctories
Vinyl Chloride (ppb)	0	2	0.5	NA	2009	ì	No D	eaching from PVC piping, ischarge from plastics ctories
Xylenes (ppm)	10	10	0.0005	NA	2009	]	No fa	ischarge from petroleum actories; Discharge from nemical factories
			Your	Sample	# Samp	les	Exceeds	10.50
Contaminants	MCLG	<u>AL</u>	Water	<u>Date</u>	Exceeding	AL	<u>AL</u>	Typical Source
Inorganic Contamin	ants							
Copper - action level at consumer taps		1.3	0.8	2012	0		No	Corrosion of household plumbing systems; Erosic of natural deposits
(ppm)  Lead - action level at consumer taps (ppb)	0	15	ţ	2012	0		No	Corrosion of household plumbing systems; Erosic of natural deposits

Term	Definition
ppm	ppm: parts per million, or milligrams per liter (mg/L)
ppb	ppb: parts per billion, or micrograms per liter (µg/L)
NA	NA: not applicable
ND	ND: Not detected
NR	NR: Monitoring not required, but recommended.

Important	Drinking Water Definitions		
	Term	Definition	ķ.
	MCLG	MCLG: Maximum Contaminant Level Goal: The level of a contaminant drinking water below which there is no known or expected risk health. MCLGs allow for a margin of safety.	ţo
	MCL	MCL: Maximum Contaminant Level: The highest level of a contam that is allowed in drinking water. MCLs are set as close to the MCL feasible using the best available treatment technology.	Gs a
	TT	TT: Treatment Technique: A required process intended to reduce level of a contaminant in drinking water.	<u> </u>
	AL	AL: Action Level: The concentration of a contaminant which, i exceeded, triggers treatment or other requirements which a water sy must follow.	f sten
Va	riances and Exemptions	Variances and Exemptions: State or EPA permission not to meet an or a treatment technique under certain conditions.	
	MRDLG	MRDLG: Maximum residual disinfection level goal. The level of drinking water disinfectant below which there is no known or experisk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.	cted

MRDL	MRDL: Maximum residual disinfectant level. The highest level of a disinfectant allowed in drinking water. There is convincing evidence the addition of a disinfectant is necessary for control of microbial contaminants.
MNR	MNR: Monitored Not Regulated
MPL	MPL: State Assigned Maximum Permissible Level

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#### Where does my water come from?

Our groundwater comes from wells using the Sparta Aquifer

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Our source water assessment has been completed and ranked moderate in terms of susceptibility and contamination. For a copy of the report please call our office at 601-992-9714.

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#### flouride

To comply with the "Regulation Governing Fluoridation of Community Water Supplies", PRVWSID-TWIN HARBOR is required to report certain results pertaining to fluoridation of our water system. The number of months in the previous calendar year in which average fluoride sample results were within the optimal range of 0.7-1.3 ppm was 12. The percentage of fluoride samples collected in the previous calendar year that was within the optimal range of 0.7-1.3 ppm was 96%.

### Results of radon monitoring

Radon is a radioactive gas that you can't see, taste, or smell. It is found throughout the U.S. Radon can move up through the ground and into a home through cracks and holes in the foundation. Radon can build up to high levels in all types of homes. Radon can also get into indoor air when released from tap water from showering, washing dishes, and other household activities. Compared to radon entering the home through soil, radon entering the home through tap water will in most cases be a small source of radon in indoor air. Radon is a known human carcinogen. Breathing air containing radon can lead to lung cancer. Drinking water containing radon may also cause increased risk of stomach cancer. If you are concerned about radon in your home, test the air in your home. Testing is inexpensive and easy. Fix your home if the level of radon in your air is 4 picocuries per liter of air (pCi/L) or higher. There are simple ways to fix a radon problem that aren't too costly. For additional information, call your state radon program or call EPA's Radon Hotline (800-SOS-RADON).

#### Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. TWIN HARBOR is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

## **Water Quality Data Table**

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of contaminants in water provided by public water systems. The table below lists all of the drinking water contaminants that we detected during the calendar year of this report. Although many more contaminants were tested, only those substances listed below were found in your water. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful in our drinking water. Removing all contaminants would be extremely expensive, and in most cases, would not provide increased protection of public health. A few naturally occurring minerals may actually improve the taste of drinking water and have nutritional value at low levels. Unless otherwise noted, the data presented in this table is from testing done in the calendar year of the report. The EPA or the State requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not vary significantly from year to year, or the system is not considered vulnerable to this type of contamination. As such, some of our data, though representative, may be more than one year old. In this table you will find terms and abbreviations that might not be familiar to you. To help you better understand these terms, we have provided the definitions below the table.

		MCLG	MCL,							
	İ	or	TT, or	Your	Ra	nge	Sample			
Contar	ninants	MRDLG	MRDL	Water	Low	<u>High</u>	<u>Date</u>	<b>Violation</b>	Typical Source	12
Disinfecta	nts & Disi	nfectant B	y-Produc	cts						
There is c	onvincing	evidence th	at additio	on of a di	sinfect	ant is n	ecessary	for control o	f microbial contaminants	)

Chlorine (as Cl2)	4	4	1.5	l	1.5	2013	No	Water additive used to control microbes
TTHMs [Total Trihalomethanes] (ppb)	NA	80	41.3	NA		2013	No	By-product of drinking water disinfection
Haloacetic Acids (HAA5) (ppb)	NA	60	37	NA	and the second trans	2013	No	By-product of drinking water chlorination
Inorganic Contamina	nts					, ,		
Nitrate [measured as Nitrogen] (ppm)	10	10	0.08	NA		2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Nitrite [measured as Nitrogen] (ppm)	1	1	0.02	ŇΑ		2013	No	Runoff from fertilizer use; Leaching from septic tanks; sewage; Erosion of natural deposits
Antimony (ppb)	6	6	0.5	NA		2010	No	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder; test addition.
Arsenic (ppb)	0	10	0.5	NA		2010	No	Erosion of natural deposits; Runoff from orchards; Runoff from glass and electronics production wastes
Barium (ppm)	2	2	0.00259 8	NA		2010	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Beryllium (ppb)	4	4	0.5	NA		2010	No	Discharge from metal refineries and coal-burning factories; Discharge from electrical, aerospace, and defense industries
Cadmium (þpb)	5	5	0.5	NA.		2010	No	Corrosion of galvanized pipes Erosion of natural deposits; Discharge from metal refineries; runoff from waste batteries and paints
Chromium (ppb)	100	100	0.636	NA		2010	No	Discharge from steel and pulp mills; Erosion of natural deposits
Cyanide [as Free Cn] (ppb)	200	200	15	NA		2010	No	Discharge from plastic and fertilizer factories; Discharge from steel/metal factories
Fluoride (ppm)	4	4	0,945	NA		2010	No	Erosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer and aluminum factories
Mercury [Inorganic] (ppb)	2	2	0.5	NA		2010	No	Erosion of natural deposits; Discharge from refineries and factories; Runoff from landfills; Runoff from cropland

Selenium (p	pb)	50	50	2.5	NA	2010	No	Discharge from petroleum and metal refineries; Erosion of natural deposits; Discharge from mines
Thallium (pj	pb)	0.5	2	0.5	NA	2010	No	Discharge from electronics, glass, and Leaching from ore-processing sites; drug factories
Volatile Or	ganie Con	taminant	s					(3)
Benzene (pp	b)	0	5	0.5	NA	2010	No	Discharge from factories; Leaching from gas storage tanks and landfills
Carbon Tetr (ppb)	achloride	0	5	0.5	NA	2010	No	Discharge from chemical plants and other industrial activities
Chlorobenze (monochloro (ppb)	!	100	100	0.5	NA	2010	No	Discharge from chemical and agricultural chemical factories
1,2-Dichlord (ppb)	ethane	0	5	0.5	NA	2010	No	Discharge from industrial chemical factories
1,1-Dichlord (ppb)	ethylene	7	7	0.5	NA	2010	No	Discharge from industrial chemical factories
cis-1,2-Dich ene (ppb)	loroethyl	70	70	0.5	NA	2010	No	Discharge from industrial chemical factories
trans-1,2-Di ylene (ppb)		100	100	0.5	NA	2010	No	Discharge from industrial chemical factories
Dichlorome (ppb)	thane	0	5	0.5	NA	2010	No	Discharge from pharmaceutical and chemical factories
1,2-Dichlord (ppb)	propane	0	5	0.5	NA	2010	No	Discharge from industrial chemical factories
Tetrachloro (ppb)	ethylene	0	5	0.5	NA	2010	No	Discharge from factories and dry cleaners
1,2,4-Trichl ne (ppb)	orobenze	70	70	0.5	NA	2010	No	Discharge from textile-finishing factories
1,1,1-Trichle (ppb)	oroethane	200	200	0.5	NA	2010	No	Discharge from metal degreasing sites and other factories
1,1,2-Trichl (ppb)	oroethane	3	5	0.5	NA	2010	No	Discharge from industrial chemical factories
Trichloroeth (ppb)	ylene	0	5	0.5	NA	2010	No	Discharge from metal degreasing sites and other factories
Toluene (pp	m)	1	1	0.5	NA	2010	No	Discharge from petroleum factories
Vinyl Chlor	ide (ppb)	0	2	0.5	NA	2010	No	Leaching from PVC piping; Discharge from plastics factories

Xylenes (ppm)	10 ·	10	0.0005	NA	2010	No	fac	scharge from petroleum tories; Discharge from emical factories
o-Dichlorobenzene (ppb)	600	600	0.5	NA	2010	No		scharge from industrial emical factories
p-Dichlorobenzene (ppb)	75	75	0.5	NA	2010	No		scharge from industrial emical factories
A Chamson militarin	The second secon		Your	Sample	# Sample	5 Ex	ceeds	
<u>Contaminants</u>	MCLG	<u>AL</u>	Water	<u>Date</u>	Exceeding A	AL .	<u>AL</u>	Typical Source
Inorganic Contami								[, i·
morganic domaini	nants							1
Copper - action leve at consumer taps (ppm)		1.3	0.2	2012	0		No	Corrosion of household plumbing systems; Erosion of natural deposits

•	
Definition	
ppm: parts per million, or milligrams per liter (mg/L)	
ppb: parts per billion, or micrograms per liter (μg/L)	1.
NA: not applicable	
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Important Prinking Water Definitio	กร	i
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TT	TT: Treatment Technique: A required process intended to reduce level of a contaminant in drinking water.	the
AŁ	AL: Action Level: The concentration of a contaminant which, i exceeded, triggers treatment or other requirements which a water sy must follow.	
Variances and Exemptions	Variances and Exemptions: State or EPA permission not to meet an or a treatment technique under certain conditions.	MCL
MRDLG	MRDLG: Maximum residual disinfection level goal. The level of drinking water disinfectant below which there is no known or experisk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.	cted
MRDL	MRDL: Maximum residual disinfectant level. The highest level of disinfectant allowed in drinking water. There is convincing evidence addition of a disinfectant is necessary for control of microbial contaminants.	e that
MNR	MNR: Monitored Not Regulated	- Ej
MPL	MPL: State Assigned Maximum Permissible Level	14

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Why are there contaminants in my drinking water?

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the Environmental Protection Agency's (EPA) Safe Drinking Water Hotline (800-426-4791).

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity: microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife; inorganic contaminants, such as salts and metals, which can be naturally occurring or result from urban stormwater runoff, industrial, or domestic wastewater discharges, oil and gas production, mining, or farming; pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses; organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems; and radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities. In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

#### How can I get involved?

We have regular board meetings on the third Thursday of each month at 9:30 A.M. or visit the website at TheRez.ms.

Water Conservation Tips

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost and no-cost ways to conserve water. Small changes can make a big difference – try one today and soon it will become second nature.

- Take short showers a 5 minute shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair and shaving and save up to 500 gallons a month.
- Use a water-efficient showerhead. They're inexpensive, easy to install, and can save you up to 750 gallons a month.
- Run your clothes washer and dishwasher only when they are full. You can save up to 1,000 gallons a month.
- Water plants only when necessary.
- Fix leaky toilets and faucets. Faucet washers are inexpensive and take only a few minutes to replace. To check your toilet for a leak, place a few drops of food coloring in the tank and wait. If it seeps into the toilet bowl without flushing, you have a leak. Fixing it or replacing it with a new, more efficient model can save up to 1,000 gallons a month.
- Adjust sprinklers so only your lawn is watered. Apply water only as fast as the soil can absorb it and during the cooler parts of the day to reduce evaporation.
- Teach your kids about water conservation to ensure a future generation that uses water wisely. Make it a family effort to reduce next month's water bill!
- Visit www.epa.gov/watersense for more information.

#### Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. LAKE HARBOR is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

## **Water Quality Data Table**

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of contaminants in water provided by public water systems. The table below lists all of the drinking water contaminants that we detected during the calendar year of this report. Although many more contaminants were tested, only those substances listed below were found in your water. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful in our drinking water. Removing all contaminants would be extremely expensive, and in most cases, would not provide increased protection of public health. A few naturally occurring minerals may actually improve the taste of drinking water and have nutritional value at low levels. Unless otherwise noted, the data presented in this table is from testing done in the calendar year of the report. The EPA or the State requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not vary significantly from year to year, or the system is not considered vulnerable to this type of contamination. As such, some of our data, though representative, may be more than one year old. In this table you will find terms and abbreviations that might not be familiar to you. To help you better understand these terms, we have provided the definitions below the table.

		=====	· · · · · · ·	1	T		1		i
		MCLG	MCL,						1 1 2
		or	TT, or	Your		nge	Sample		-
<u>Contami</u>	<u>nants</u>	MRDLG	MRDL	Water	Low	<u>High</u>	<u>Date</u>	Violation	Typical Source
Disinfectant								·	
(There is con	vincing o	evidence th	at additio	n of a di	sinfect	ant is n	ecessary i	for control o	f microbial contaminants)
Chlorine (as (ppm)		4	4	1.2	0.9	1.4	2013	No	Water additive used to control microbes
TTHMs [Tot Trihalometh (ppb)		NA	80	10.81	NA		2011	No	By-product of drinking water disinfection
Inorganic C	ontamin	ants				and and desired the second	åi di kreedi (kaliiki ja di esterre en en id		:
Barium (ppn	n)	2	2	0.0036	0.001	0.0036	2013	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Fluoride (ppr	n)	4	4	0.66	0.23	0.66	2013	No	Erosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer and aluminum factories
Nitrate [meas Nitrogen] (p)		10	10	0.08	0.08	0.08	2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Nitrite [meas Nitrogen] (pp		1	1	0.02	0.02	0.02	2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Antimony (p	pb)	6	6	0.5	0.5	0.5	2013	No	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder; test addition.
Arsenic (ppb	)	0	10	0.5	0.5	0.5	2013	No	Erosion of natural deposits; Runoff from orchards; Runoff from glass and electronics production wastes

Beryllium (pp	b)	4	4	0.5	0.5	0.5	2013	No	Discharge from metal refineries and coal-burning factories; Discharge from electrical, aerospace, and defense industries
Cadmium (ppl	b)	5	5	0.5	0.5	0.5	2013	No	Corrosion of galvanized pipe Erosion of natural deposits; Discharge from metal refineries; runoff from waste batteries and paints
Chromium (pr	ob)	100	100	0.7	0.7	I	2013	No	Discharge from steel and pul mills; Erosion of natural deposits
Cyanide [as Fi (ppb)	rec Cn]	200	200	15	15	15	2013	No	Discharge from plastic and fertilizer factories; Discharge from steel/metal factories
Mercury [Inor (ppb)	ganic]	2	2	0.5	0.5	0.5	2013	No	Erosion of natural deposits; Discharge from refineries an factories; Runoff from landfills; Runoff from cropland
Selenium (ppb	))	50	50	2.5	2.5	2.5	2013	No	Discharge from petroleum ar metal refineries; Erosion of natural deposits; Discharge from mines
Thallium (ppb	))	0.5	2	0.5	0.5	0.5	2013	No	Discharge from electronics, glass, and Leaching from ore processing sites; drug factories
Volatile Orga	nie Con	taminant	S						
Toluene (ppm	)	1	1	0.5	NA		2010	No	Discharge from petroleum factories
Xylenes (ppm	)	10	10	0.0005	NA		2010	No	Discharge from petroleum factories; Discharge from chemical factories
Carbon Tetrac (ppb)	chloride	0	5	0.5	NA		2010	No	Discharge from chemical plants and other industrial activities
Chlorobenzen (monochlorob (ppb)		100	100	0.5	NA	·	2010	No	Discharge from chemical and agricultural chemical factorie
o-Dichloroben (ppb)	ухеле	600	600	0.5	NA		2010	No	Discharge from industrial chemical factories
p-Dichloroben (ppb)		75	75	0.5	NA		2010	No	Discharge from industrial chemical factories
l,2-Dichloroe (ppb)		0	5	0.5	NA		2010	No	Discharge from industrial chemical factories
1,1-Dichlorce (ppb)	thyleпе	7	7	0,5	NA		2010	No	Discharge from industrial chemical factories
cis-1,2- Dichloroethyle (ppb)	ene	70	70	0.5	NA		2010	No	Discharge from industrial chemical factories

trans-1,2- Dichloroethy (ppb)	lene	100	100	0.5	NA	. 20	10	No	Discharge from industrial chemical factories
Dichloromet (ppb)	hane	0	5	0.5	NA	20	)10	No	Discharge from pharmaceutical and chemical factories
1,2-Dichlord (ppb)	propane	0	5	0.5	NA	20	)10	No	Discharge from industrial chemical factories
Ethylbenzen	e (ppb)	700	700	0.5	ÑΑ	20	010	No	Discharge from petroleum refineries
Styrene (ppl	<b>)</b> )	100	100	0.5	NA	20	010	No	Discharge from rubber and plastic factories; Leaching from landfills
Tetrachloroe (ppb)	thylene	0	5	0.5	NA	2(	010	No	Discharge from factories and dry cleaners
1,2,4- Trichlorober (ppb)	nzene	70	70	0.5	NA	20	010	No	Discharge from textile- finishing factories
1,1,1-Trichl (ppb)	oroethane	200	200	0.5	NA	20	010	No	Discharge from metal degreasing sites and other factories
1,1,2-Trichl (ppb)	oroethane	3	5	0.5	NA	20	010	No	Discharge from industrial chemical factories
Trichloroeth (ppb)	ylene	0	5	0.5	NA	20	010	No	Discharge from metal degreasing sites and other factories
Vinyl Chlor	ide (ppb)	0	2	0.5	NA	20	010	No	Leaching from PVC piping; Discharge from plastics factories

Unit Descriptions		1:
Term	Definition	1:1
ppm	ppm: parts per million, or milligrams per liter (mg/L)	į
ppb	ppb: parts per billion, or micrograms per liter (μg/L)	
NA	NA: not applicable	- Colombia
ND	ND: Not detected	:
NR	NR: Monitoring not required, but recommended.	1

Important Drinking Water Defin	nitions
Term	Definition
MCLG	MCLG: Maximum Contaminant Level Goal: The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.
MCL	MCL: Maximum Contaminant Level. The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.
TT	TT: Treatment Technique: A required process intended to reduce the level of a contaminant in drinking water.

AL	AL: Action Level: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements which a water system must follow.
Variances and Exemptions	Variances and Exemptions: State or EPA permission not to meet an MC or a treatment technique under certain conditions.
MRDLG	MRDLG: Maximum residual disinfection level goal. The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.
MRDL	MRDL: Maximum residual disinfectant level. The highest level of a disinfectant allowed in drinking water. There is convincing evidence the addition of a disinfectant is necessary for control of microbial contaminants.
MNR	MNR: Monitored Not Regulated
MPL	MPL: State Assigned Maximum Permissible Level

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#### Is my water safe?

We are pleased to present this year's Annual Water Quality Report (Consumer Confidence Report) as required by the Safe Drinking Water Act (SDWA). This report is designed to provide details about where your water comes from, what it contains, and how it compares to standards set by regulatory agencies. This report is a snapshot of last year's water quality. We are committed to providing you with information because informed customers are our best allies.

#### Do I need to take special precautions?

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the \$afe Water Drinking Hotline (800-426-4791).

#### Where does my water come from?

Our groundwater is from the Cockfield Formation and Sparta Aquifer.

#### Source water assessment and its availability

Our source water assessment has been done and we ranked MODERATE in terms of susceptibility to contamination. For a copy of the report please call our office at 601-992-9714

#### Why are there contaminants in my drinking water?

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agriculture, urban stormwater runoff, and residential uses; organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems; and radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities. In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

#### How can I get involved?

Pearl River Valley Water Supply has regular board meetings on the third Thursday of each month or call 601-992-9714

#### Description of Water Treatment Process

Your water is treated by disinfection. Disinfection involves the addition of chlorine or other disinfectant to kill dangerous bacteria and microorganisms that may be in the water. Disinfection is considered to be one of the major public health advances of the 20th century.

#### Water Conservation Tips

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost and no-cost ways to conserve water. Small changes can make a big difference – try one today and soon it will become second nature.

- Take short showers a 5 minute shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair and shaving and save up to 500 gallons a month.
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  absorb it and during the cooler parts of the day to reduce evaporation.
- Teach your kids about water conservation to ensure a future generation that uses water wisely. Make it a family effort to reduce next month's water bill!
- Visit www.epa.gov/watersense for more information.

#### Source Water Protection Tips

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source in several ways:

- Eliminate excess use of lawn and garden fertilizers and pesticides they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain your system to reduce leaching to water sources or consider connecting to a public water system.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Volunteer in your community. Find a watershed or wellhead protection organization in your community and volunteer to help. If there are no active groups, consider starting one. Use EPA's Adopt Your Watershed to locate groups in your community, or visit the Watershed Information Network's How to Start a Watershed Team.
- Organize a storm drain stenciling project with your local government or water supplier.
   Stencil a message next to the street drain reminding people "Dump No Waste Drains to River" or "Protect Your Water." Produce and distribute a flyer for households to remind residents that storm drains dump directly into your local water body.

#### FLOURIDE

To comply with the "Regulation Governing Fluoridation of Community Water Supplies", PRVW\$D-PELAHATCHIE BAY is required to report certain results pertaining to fluoridation of our water system. The number of months in the previous calendar year in which average fluoride sample results were within the optimal range of 0.7-1.3 ppm was 11. The percentage of fluoride samples collected in the previous calendar year that was within the optimal range of 0.7-1.3 ppm was 81%.

#### Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. PRVWSD PELEHATCHIE BAY is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

## **Water Quality Data Table**

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of contaminants in water provided by public water systems. The table below lists all of the drinking water contaminants that we detected

during the calendar year of this report. Although many more contaminants were tested, only those substances listed below were found in your water. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful in our drinking water. Removing all contaminants would be extremely expensive, and in most cases, would not provide increased protection of public health. A few naturally occurring minerals may actually improve the taste of drinking water and have nutritional value at low levels. Unless otherwise noted, the data presented in this table is from testing done in the calendar year of the report. The EPA or the State requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not vary significantly from year to year, or the system is not considered vulnerable to this type of contamination. As such, some of our data, though representative, may be more than one year old. In this table you will find terms and abbreviations that might not be familiar to you. To help you better understand these terms, we have provided the definitions below the table.

								***************************************		
	]	MCLG	MCL,				_			
		or	TT, or	Your	ι,	nge	Sample	ł		
Conta	minants	MRDLG	MRDL	<u>Water</u>	Low	High	<u>Date</u>	<u>Violation</u>	Typical Source	Pall N
	ants & Disir				zangan apakan apa	tta-poptentananana				2000
(There is	convincing e	vidence tha	t addition	of a disi	nfectar	it is nec	essary fo	r control of	nicrobial contaminants)	
Chlorine (ppm)	(as Cl2)	4	4	1.2	8.0	1.8	2013	No	Water additive used to control microbes	) L
TTHMs Trihalom (ppb)	4	NA	80	16.7	12.15	16.7	2013	No	By-product of drinking water disinfection	r
Haloacet (HAA5)	1	NA	60	15	10	15	2013	No	By-product of drinking water chlorination	Ļ.
Inorgani	c Contamin	ants								
Cyanide (ppb)	[as Free Cn]	200	200	15	15	15	2013	No	Discharge from plastic and fertilizer factories; Discharge from steel/metal factories	3
Antimon	y (ppb)	6	6	0.5	0.5	0.5	2013	No	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder test addition.	;
Arsenic (	ppb)	0	10	0.5	0.5	0.5	2013	No	Erosion of natural deposits; Runoff from orchards; Runo from glass and electronics production wastes	ff
Barium (	ppm)	2	2	0.003	0.001 9	0.003	2013	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits	
Nitrate [i Nitrogen	neasured as ] (ppm)	10	10	0.08	0.08	0,08	2013	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits	
Fluoride	(ppm)	4	4	0.721	0.34	0.721	2013	No	Brosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer and aluminum factories	1

Beryllium	(ррь)	4	4	0,5	0.5	0.5	2013	No	Discharge from metal refineries and coal-burning factories; Discharge from electrical, acrospace, and defense industries
Cadmium	(ppb)	5	5	0.5	0.5	0.5	2013	No	Corrosion of galvanized pipes; Erosion of natural deposits; Discharge from metal refineries; runoff from waste batteries and paints
Chromiur	n (ppb)	100	100	0.5	0.5	0.5	2013	No	Discharge from steel and pulp mills; Erosion of natural deposits
Mercury (ppb)	[Inorganic]	2	2	0.5	0.5	0.5	2013	No	Erosion of natural deposits; Discharge from refineries and factories; Runoff from landfills; Runoff from cropland
Selenium	(ppb)	50	50	2.5	2.5	2.5	2013	No	Discharge from petroleum and metal refineries; Erosion of natural deposits; Discharge from mines
Volatile	Organic Con	taminant	3				4.5.4		
Benzene	(ppb)	0	5	0.5	NA	1	2012	No	Discharge from factories; Leaching from gas storage tanks and landfills
Carbon T (ppb)	etrachloride	0	5	0.5	NA.		2012	No	Discharge from chemical plants and other industrial activities
Chlorobe (monoch (ppb)	nzene orobenzene)	100	100	0.5	NA		2012	No	Discharge from chemical and agricultural chemical factories
o-Dichlo (ppb)		600	600	0.5	NA		2010	No	Discharge from industrial chemical factories
(ppb)	obenzene	75	75	0.5	NA		2012	No	Discharge from industrial chemical factories
1,2-Dich (ppb)	oroethane	0	5	0.5	NA		2012	No	Discharge from industrial chemical factories
I,I-Dichl (ppb)	oroethylene	7	7	0.5	NA		2012	Nο	Discharge from industrial chemical factories
cis=1,2= Dichloro (ppb)		70	70	0.5	NA		2010	No	Discharge from industrial chemical factories
trans-1,2 Dichloroe (ppb)		100	100	0.5	NA		2012	No	Discharge from industrial chemical factories
Dichloror (ppb)	nethane	0	5	0.5	NA	L	2012	No	Discharge from pharmaceutical and chemical factories
1,2-Dichl (ppb)	oropropane	0	5	0.5	NA		2012	No	Discharge from industrial chemical factories

Ethylbenze	ne (ppb)	700	700	0.5	NA		2012	1		Discharge from petroleum refineries	
Styrene (pp	ob)	100	100	0.5	NA		2012	]	No	Discharge from rubber and plastic factories; Leaching from landfills	
Tetrachloro (ppb)	ethylene	0	5	0.5	NA		2012	]	Nο	Discharge from factories and dry cleaners	1
1,2,4- Trichlorobe (ppb)	enzene	70	70	0.5	NA		2012	<u>.</u> !	No	Discharge from textile- finishing factories	
	loroethane	200	200	0.5	NA <sup>,</sup>		2012		No	Discharge from metal degreasing sites and other factories	:
1,1,2-Trich (ppb)	loroethane	3	5	0.5	NA		2012		No	Discharge from industrial chemical factories	
Trichloroet (ppb)	thylene	0	5	0.5	NA		2012		No	Discharge from metal degreasing sites and other factories	
Toluene (p	pm)	1	1	0.0005	NA		2012		No	Discharge from petroleum factories	
Vinyl Chlo	oride (ppb)	0	2	0.5	NA	·	2012	-	No	Leaching from PVC piping; Discharge from plastics factories	
Xylenes (p	pm)	10	10	0.0005	NA		2012		No	Discharge from petroleum factories; Discharge from chemical factories	
	inchesional de sin lore			Your	Sam	ple	# Sampl	es	Excee	eds	
Conten	<u>ninants</u>	MCLG	<u>al</u>	Water	Da	te	Exceeding	AL	AL	Typical Source	i
Inorganic	Contamina	ints			1				·		<u> </u>
Lead - actio		0	15	3	20	10	0		No	Corrosion of household plumbing systems; Erosi of natural deposits	on
Copper - ac at consume (ppm)		1.3	1.3	0.4	20	11	0		No	Corrosion of household plumbing systems; Erosi of natural deposits	on

Unit Des	criptions		
	Term	Definition	1
	ррт	ppm: parts per million, or milligrams per liter (mg/L)	
	ррь	ppb: parts per billion, or micrograms per liter (μg/L)	
	NA	NA: not applicable	
	ND	ND: Not detected	
	NR	NR: Monitoring not required, but recommended.	

Importa	nt Drinking Water Definitions		
,	Terni	Definition	
	MCLG	MCLG: Maximum Contaminant Level Goal: The level of a contamina in drinking water below which there is no known or expected risk to	
	ì	health. MCLGs allow for a margin of safety.	

MCL	MCL: Maximum Contaminant Level: The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.
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Variances and Exemptions	Variances and Exemptions: State or EPA permission not to meet an MCL or a treatment technique under certain conditions.
MRDLG	MRDLG: Maximum residual disinfection level goal. The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.
MRDL	MRDL: Maximum residual disinfectant level. The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.
MNR	MNR: Monitored Not Regulated
MPL	MPL: State Assigned Maximum Permissible Level

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